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## Nucleosides, Nucleotides and Nucleic Acids

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### SYNTHESIS AND BASE PAIRING PROPERTIES OF 9-DEAZAPURINE N<sup>7</sup>-NUCLEOSIDES IN OLIGONUCLEOTIDE DUPLEXES AND TRIPLEXES

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## SYNTHESIS AND BASE PAIRING PROPERTIES OF 9-DEAZAPURINE N<sup>7</sup>-NUCLEOSIDES IN OLIGONUCLEOTIDE DUPLEXES AND TRIPLEXES

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### ABSTRACT

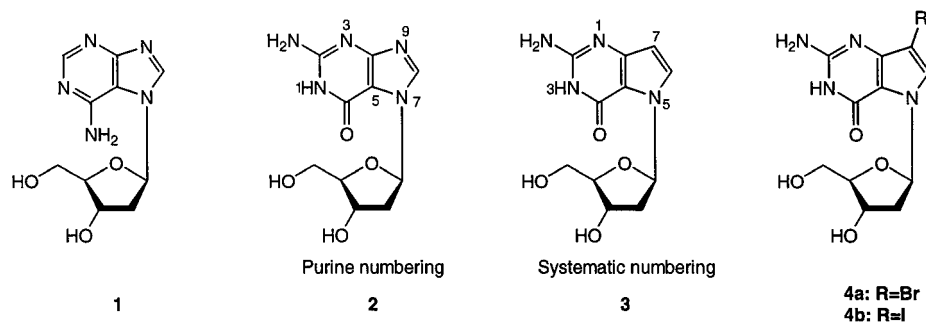
The 9-deazaguanine N<sup>7</sup>-2'-deoxyribofuranoside (**3**) as well as the bromo and iodo derivatives **4a,b** were synthesized and incorporated in oligonucleotide duplexes and triplexes. Their base pairing properties were investigated and compared with those of the parent purine N<sup>7</sup>-2'-deoxyribofuranosides.

### INTRODUCTION

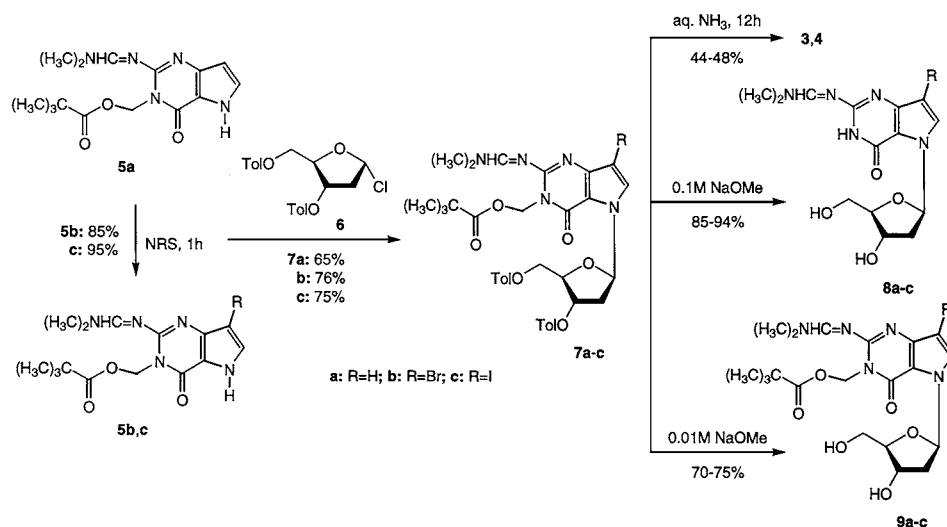
Several reports have communicated on the base pairing properties of nucleosides linked via nitrogen-7 or position-8 to the nucleobase (1-3). The nucleosides **1** and **2** with the unusual N<sup>7</sup>-glycosylation site form stable base pairs in duplex and triplex DNA (1,2,4). Now, we report on the synthesis and base pairing properties of the 9-deazaguanine N<sup>7</sup>-nucleosides **3** and **4a,b** which are remarkably stable analogs of the purine nucleosides. The pyrrolo[3,2-d]pyrimidine nucleosides were incorporated in oligonucleotide duplexes as well as in triplex-DNA. Their base pairing properties were compared with compound **2**.

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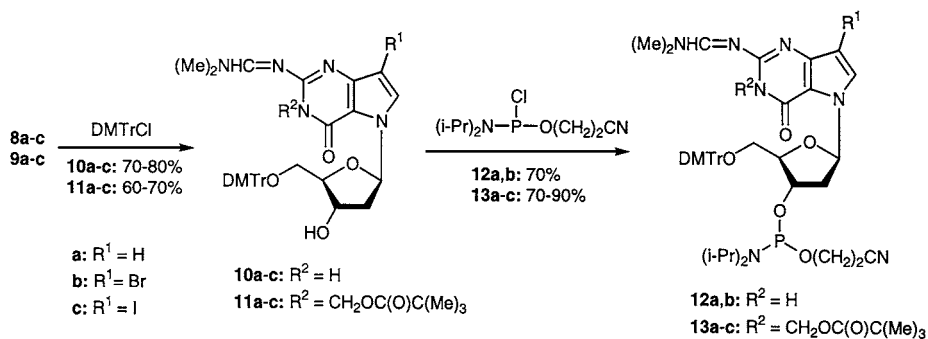
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Scheme 1.



Scheme 2.



Scheme 3.



**Table 1.** Antiparallel Duplexes with N<sup>7</sup>G<sub>d</sub>-iC<sub>d</sub> and c<sup>9</sup>N<sup>7</sup>G<sub>d</sub>-iC<sub>d</sub> Base Pairs<sup>a)</sup>

aps-Duplexes	T <sub>m</sub> [°C]	aps-Duplexes	T <sub>m</sub> [°C]
5'-d(TAGGTCAATACT)	<b>14</b>	5'-d(TA <b>4a4a</b> T CAATA CT)	<b>16</b>
3'-d(ATCCAGTTATGA)	<b>15</b>	3'-d(ATiCiC A GTTATGA)	<b>17</b>
5'-d(TA <b>iCiCT</b> CAATA CT)	<b>18</b>	5'-d(TA <b>4b4b</b> TCAATA CT)	<b>20</b>
3'-d(AT <b>2 2</b> AGTTATGA)	<b>19</b>	3'-d(ATiCiC A GTTATGA)	<b>17</b>
5'-d(TA <b>3 3</b> T CAATACT)	<b>21</b>		
3'-d(ATiCiC A GTTATGA)	<b>17</b>		

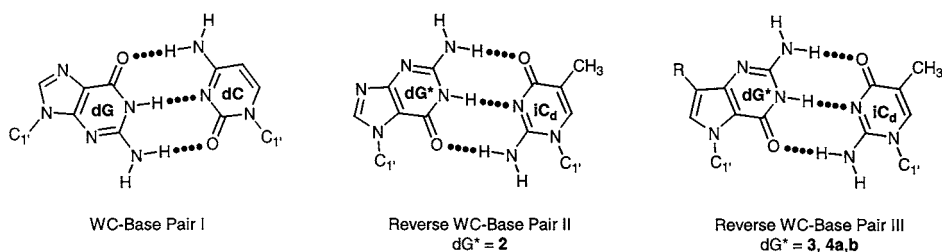
<sup>a)</sup> Measured at 260 nm in 0.1M NaCl, 10 mM MgCl<sub>2</sub>, and 10 mM Na-cacodylate (pH 7.0) at 5 μM of single strand concentration. iC<sub>d</sub> = 5-methyldeoxyisocytidine.

## RESULTS AND DISCUSSION

The synthesis of the nucleosides **3** (**5**) and **4a,b** utilized the nucleobases **5a** (**6**) and **5b–c** as precursors. Compounds **5b,c** were obtained from **5a** by regioselective halogenation. The stereoselective nucleobase-anion glycosylation of **5a–c** with the halogenose **6** furnished the N<sup>7</sup>-β-D-nucleosides **7a–c**. Deblocking of **7a–c** (0.01M NaOMe) resulted in a selective detoluoylation (→**9a–c**), while treatment of **7a–c** with 0.1M NaOMe cleaved the pivaloyloxymethyl group (→**8a–c**). Treatment of **7a–c** with conc. aq. NH<sub>3</sub> furnished the nucleosides **3** and **4**.

Next, formamidine protecting groups were introduced. The half-life values of depotection were 8 min for **8a**, 12 min (**8b**) and 17 min (**8c**) (25% aq. NH<sub>3</sub>, 40°C). The protected nucleosides **8a,b** as well as **9a–c** were tritylated (→**10a–c** and **11a–c**) and were converted in building blocks for solid-phase synthesis (**12a,b** and **13a–c**).

The base-pairing properties of the nucleosides **3** and **4a,b** were investigated on duplexes with antiparallel chain orientation. According to Table 1 compounds **3** and **4a,b** form base pairs with 5-methyl-2'-deoxyisocytidine (iC<sub>d</sub>) in aps-duplexes (**16·17**, **20·17** and **21·17**). This is in line with observations made on the base pairing properties of the purine N<sup>7</sup>-nucleoside **2** (**2**) (see duplex **18·19**). In this case a strong base pair with iC<sub>d</sub> was formed. Furthermore, the bulky bromo and iodo substituents



**Scheme 4.**



**Table 2.** Oligonucleotides Forming the Base Triplets 3:dG-dC and dC:dG-dC<sup>a)</sup>

5'-d(TTTXTTTTXXXTT 5'-d(GCGCGAAAGAAAAGAGAGAACCCGG) 3'-d(CGCGCTTCTTTTCTCTTGGGCC)			5'-d(TTTXTTTTXXXXXTT 5'-d(GCGCGAAAGAAAAGGGGGAACCCGG) 3'-d(CGCGCTTCTTTTCCCCCTTGGGCC)		
X	T <sub>m</sub> [°C] pH 6.5	T <sub>m</sub> [°C] pH 8.0	X	T <sub>m</sub> [°C] pH 6.5	T <sub>m</sub> [°C] pH 8.0
3	45/72	44/72	3	53/74	52/74
C	45/72	16/72	C	24/74	n.t./74

<sup>a)</sup> Measured at 260 nm in 10 mM HEPES, 50 mM NaCl, 10 mM MgCl<sub>2</sub> and 0.5 mM spermine at 1 μM of single strand concentration. N.t.: no transition.

introduced in position-9 of the heterocycle (**4a,b**) are well accommodated in the duplex structure.

The results of Table 1 raise the question about the structure of the base pairs. Several base pair motives were considered. Due to the almost identical T<sub>m</sub>-values of the duplexes containing **2,3** and **4a,b** and the fact that the tridentate base pairs (II and III) are similar to the tridentate WC-base pair of I, the motifs II and III are the most likely ones.

Earlier, the triplex formation of oligomers containing N<sup>7</sup>G<sub>d</sub>(**2**) as well as of an acyclic derivative of **2** recognizing a dG-dC pair of a duplex DNA was described (4,7). Now, compound **3** was introduced into the third strand of a triplex (Table 2). Sigmoidal melting curves with a biphasic profile were obtained in all cases. The higher T<sub>m</sub>-values are assigned to the duplex melting while the lower ones describe the triplex dissociation.

Table 2 shows that triplex melting is independent from the pH-value, when compound **3** is a constituent of the base triplet. The triplexes containing dC-residues at that position require protonation before triplex formation. As the sugar phosphate backbone of the pyrimidine-rich third strand becomes distorted by the incorporation of **3**, the stability of the triplex is higher when the **3**-residues are in a consecutive order instead of being dispersed. The advantage of compound **3** compared to **2** is its remarkable glycosylic bond stability and the possibility to introduce reporter groups at the 9-position (**4a,b**).

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